NOTES

A Novel Neuronal Cell Protecting Substance, Halxazone, Produced by Streptomyces halstedii

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Parkinson's disease (PD) is characterized by selective degeneration of dopaminergic neurons in the substantia nigra. Although the accurate etiology of Parkinson's disease is not clear, one of the hypothetical factors is oxidative stress. The excess administration of L-DOPA, which is an effective drug in PD therapy, results in the worsening of Parkinson's disease, by generating reactive oxygen species $(ROS)^{1\sim3}$. Thus, the advance of PD may be expected to be overcome by substances which suppress L-DOPA toxicity.

In the course of our screening program for substances which protect rat pheochromocytoma PC12 cells from L-DOPA toxicity, we isolated a novel compound designated as halxazone (8-hydoroxymethyl-1-methoxy-3*H*-phenoxazin-3-one) (1) from an actinomycete identified as *Streptomyces halstedii* 4029-SVS1. We report herein the isolation, structure elucidation, and biological activities of 1.

The producing microorganism, *Streptomyces halstedii* 4029-SVS1, which was isolated from a soil sample collected in Takehara, Hiroshima Prefecture, Japan, was cultivated at 27°C in a 60-liter jar fermenter containing 30 liters of a medium consisting of 2.0% glycerol, 1.0% molasses, 0.5% casein, 0.1% polypepton and 0.4% CaCO₃ (pH 7.2) with the agitation rate of 400 rpm and aeration at 30 liters per minute. After fermentation for 72 hours, the culture broth was centrifuged to give a supernatant, which was extracted with an equal volume of EtOAc twice. The extract was dried over Na₂SO₄ and concentrated *in vacuo* to dryness to yield a dark brown oil. The oily material was subjected to silica gel column chromatography developed

with a solvent system of $CHCl_3$ -MeOH (40:1). The active eluate was then applied to Toyopearl HW-40F and developed with MeOH. The active fraction was purified by MPLC using a Senshu Pak PEGASIL ODS column (37 mm i.d.×300 mm) with 40% MeOH. Finally, the pure sample of 1 (3.9 mg) was obtained by HPLC using a Senshu Pak PEGASIL ODS column (20 mm i.d.×250 mm) with 35% MeOH as a congener of michigazone (2)⁴ and 4demethoxymichigazone (3)⁵.

The IR absorption of 1 (mp. $201 \sim 202^{\circ}$ C) at 3395 and 1640 cm^{-1} indicated the presence of a hydroxyl and a ketone function, respectively. The UV and visible absorptions of 1 (λ_{max} nm (ε) in MeOH: 204 (18,300) 254 (13,500) 365 (10,000) 456 (4,700)) suggested that 1 possesses a phenoxazone nucleus^{4~6)}. The molecular formula of 1 was established as C₁₄H₁₁NO₄ by HRFAB-MS (*m*/z 258.0793 (M+H)⁺, calcd 258.0766).

The ¹H and ¹³C NMR data of **1** exhibited 14 carbon and 11 proton signals. These ¹H and ¹³C assignments are revealed by a heteronuclear multiple-quantum coherence (HMQC) and summarized in Table 1. The HMBC spectra of **1** together with proton spin couplings between 6-H ($\delta_{\rm H}$ 7.29), 7-H ($\delta_{\rm H}$ 7.57) and 9-H ($\delta_{\rm H}$ 7.84) revealed the presence of a 1,2,4-trisubstituted benzene ring. A singlet hydroxymethyl proton signal 11-H ($\delta_{\rm H}$ 4.67) showed ¹H-¹³C long-range couplings to C-7 ($\delta_{\rm C}$ 132.7), C-8 ($\delta_{\rm C}$ 140.0) and C-9 ($\delta_{\rm C}$ 128.9) in the HMBC spectra. Thus, the hydroxymethyl residue was located at the C-8 position. Other two quaternary carbons in the trisubstituted benzene

Fig. 1. Structure of halxazone (1), michigazone (2) and 4-demethoxymichigazone (3).



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$\delta_{\rm C}$	$\delta_{\rm H}$ (multiplicity, $J = {\rm Hz}$)
159.2	
108.0	6.08 (dJ = 2.0 Hz)
187.3	
104.8	6.12 (d J = 2.0 Hz)
149.6	
143.9	
116.4	7.29 (d $J = 8.5$ Hz)
132.7	7.57 (dd J = 2.0, 8.5 Hz)
140.0	
128.9	7.84 (d $J = 2.0$ Hz)
132.4	
144.6	
63.4	4.67 (s 2H)
57.1	3.95 (s 3H)
	$\frac{\delta_{\rm C}}{159.2}$ 108.0 187.3 104.8 149.6 143.9 116.4 132.7 140.0 128.9 132.4 144.6 63.4 57.1

Table 1. ${}^{13}C$ and ${}^{1}H$ NMR assignments of 1.

¹H and ¹³C NMR spectra were measured at 500 MHz and 125 MHz, respectively, in $CDCl_3 + CD_3OD$ (5 : 1)

ring, C-5a ($\delta_{\rm C}$ 143.9) and C-9a ($\delta_{\rm C}$ 132.4) were assigned by ¹H-¹³C long-range couplings as shown in Fig. 2. Based on the phenoxazone chromophore deduced by UV spectral data *vide supra*, the relationship between two isolated *sp*² protons 2-H ($\delta_{\rm H}$ 6.08) and 4-H ($\delta_{\rm H}$ 6.12) were assigned to the *meta* position from the result of the coupling constant ($J_{\rm H-H}$ =2.0 Hz) as shown in Fig. 2. In the HMBC spectra of 1, 2-H and 4-H were strongly *meta*-coupled to a quaternary carbon C-10a ($\delta_{\rm C}$ 144.6), and were *ortho*-coupled to a carbonyl carbon C-3 ($\delta_{\rm C}$ 187.3). In addition, 2-H also longrange coupled to a quaternary carbon C-1 ($\delta_{\rm C}$ 159.2), which in turn long-range coupled to a methoxy proton ($\delta_{\rm H}$ 3.95) revealing that the methoxy residue was substituted at C-1. An *ortho*-coupling from 4-H to C-4a ($\delta_{\rm C}$ 149.6) established the assignments of all ¹³C and ¹H signals.

Taking into consideration the presence of the phenoxazone chromophore and molecular formula of 1, C-10a was deduced to be substituted by a nitrogen atom, and the remaining oxygen atom was linked to C-4a. Thus, the conjugated system consisting of these units was established as shown in Fig. 2. Two possibilities were considered about the connectivity between ring A and ring B. Due to a low field shift of 9-H ($\delta_{\rm H}$ 7.84) compared with two other aromatic protons 6-H ($\delta_{\rm H}$ 7.29) and 7-H ($\delta_{\rm H}$ 7.57) in the

• Fig. 2. Analyses of ¹³C-¹H long-range couplings in HMBC spectrum of 1.



same ring system, 9-H must be at a *peri* position to the nitrogen atom^{7,8)}. The ¹³C chemical shifts of C-5a and C-9a also supported the connectivity between ring A and ring B as shown in Fig. 2.

The protective effect of 1, 2 and 3 on L-DOPA toxicity in PC12 cells was examined employing the method of MIGHELI et al¹⁾ after minor modification. 1, 2 and 3 potentially protected PC12 cells from L-DOPA (50 μ M) toxicity with EC_{50} values of 15.4, 10.7 and 30.0 nm, respectively. Since 3 was reported as the neuroprotecting substance which suppressed L-glutamate toxicity in N18-RE-105 cells^{9,10}, the protective effect of 1 against Lglutamate toxicity was observed. 1 also protected N18-RE-105 cells from L-glutamate toxicity with EC₅₀ value of 19.1 nm, which was almost identical to that of 2. L-Glutamate toxicity in N18-RE-105 cells was considered to be induced by the accumulation of $ROS^{11\sim13}$. Thus, 1 was deduced to protect PC12 cells from L-DOPA toxicity by exhibiting anti-ROS activity. Detailed studies on biological activities are now under way.

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